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# Molecular Crystals and Liquid Crystals

Publication details, including instructions for authors and subscription information: <a href="http://www.tandfonline.com/loi/gmcl16">http://www.tandfonline.com/loi/gmcl16</a>

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To cite this article: K. P.J. Williams , D. L. Gerrard , D. C. Bott & C. K. Chai (1985): Resonance Raman Studies of the Elimination and Transformation Reaction of Durham Polyacetylene, Molecular Crystals and Liquid Crystals, 117:1, 23-28

To link to this article: <a href="http://dx.doi.org/10.1080/00268948508074588">http://dx.doi.org/10.1080/00268948508074588</a>

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Mol. Cryst. Liq. Cryst. 1985, Vol. 117, pp. 23-28 0026-8941/85/1174-0023/\$15.00/0 © 1985 Gordon and Breach, Science Publishers, Inc. and OPA Ltd. Printed in the United States of America

RESONANCE RAMAN STUDIES OF THE ELIMINATION AND TRANSFORMATION REACTION OF DURHAM POLYACETYLENE

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Abstract The elimination and transformation reaction of. Durham polyacetylene has been monitored by resonance Raman spectroscopy. The resonance Raman and photoluminescence spectrum from short <u>cis</u> segments, present in the pre-cursor polymer, has been recorded. A distribution of <u>cis</u> polyacetylene sequence lengths has been observed from Raman measurements made at different time intervals as the reaction proceeds, using  $\lambda_{\rm e}$  647.lnm. The  $\nu_{\rm e}$  values recorded gradually decrease as the reaction proceeds the limiting value of 1472 cm<sup>-1</sup> is significantly less than the 1450-1460 cm<sup>-1</sup> values reported for the Shirakawa polymer.

#### INTRODUCTION

Most of the published Raman spectroscopic data on polyacetylene relate to the Shirakawa type of material. Our investigations have been, in the main, into the Durham material. The Durham route to polyacetylene is unique in that the polyacetylene is a final product from an elimination and transformation reaction (Figure 1).

This study details the time - resolved resonance Raman data recorded as the elimination and transformation reaction proceeds. The reaction, performed at three temperatures, ambient, 65 and 85°C has been monitored by observing the spectral changes that occur with time. The structural implications from these observations are

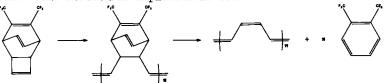


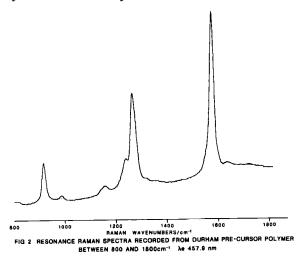
FIG 1 EDWARDS AND FEAST PREPARATION OF POLYACETYLENE

discussed.

#### RESULTS AND DISCUSSION

#### Pre-cursor Polymer

Figure 2 shows the resonance Raman spectrum recorded at 77K, from the Durham pre-cursor, using blue ( $\lambda_e$  457.9nm) laser excitation. It is apparent from the figure that the <u>trans</u> polyacetylene content of the sample is low. The position of the three prominent bands is



914, 1261 and 1565 cm<sup>-1</sup>, respectively. It appears that the Raman spectrum recorded from the pre-cursor polymer can reasonably be assigned to some <u>cis</u> polyacetylene that is present. This assignment is further supported by the presence of a long overtone/combination progression and a photoluminescence band centred at 2.3eV. A comparison of Raman data recorded from Shirakawa <u>cis</u> polyacetylene and from the Durham pre-cursor polymer is given in Table 1. It can be seen from Table 1 that significant differences exist between the two sets of data. The wavenumber shifts are thought to reflect a difference in <u>cis</u> polyacetylene chain length. The Raman spectrum obtained from the pre-cursor polymer, using  $\lambda_{\rm e}$  647.1nm was much weaker, since resonance conditions were far from optimum, however, the wavenumber values recorded were identical.

The difference in the wavenumber, and photoluminescence band, values given in Table 1 coupled with the lack of dispersion of the Raman bands when changing from blue to red excitation, is attributed to the presence of short <u>cis</u> polyacetylene segments in the pre-cursor polymer. These segments are thought to be significantly shorter than those previously observed from Shirakawa type polymer. However, the existence of short <u>cis</u> polyacetylene segments, in the pre-cursor, is not unreasonable if some elimination and transformation has occurred prior to analysis.

TABLE 1 Raman wavenumbers (cm<sup>-1</sup>) recorded from Durham pre-cursor polymer at 77K and Shirakawa <u>cis</u> polyacetylene at 77K between 800 and 1800 cm<sup>-1</sup>

Durham Pre-cursor Polymer <sup>1</sup> Wavenumber <del>s</del> cm <sup>-</sup> 1	Shirakawa <u>cis</u> Polyacetylene <sup>2</sup> Wavenumbers/cm <sup>-1</sup>		
805	825		
914	910		
984	980		
1133*	1112*		
1148*	_		
1230	_		
1261	1250		
1316	-		
-	1497*		
1565	1541		
1617	_		

Photoluminescence centred at √2.3eV

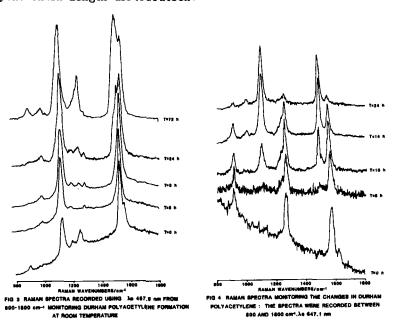
### 2. Elimination and Transformation Reaction

Figures 3 and 4 show the Raman spectra recorded at various time intervals, using  $\lambda_e$  457.9 and 647.1nm, respectively, as the reaction proceeds at ambient temperature. Further details of the

<sup>2.</sup> Photoluminescence centred at ~1.9eV

<sup>\*</sup> Indicates trans polyacetylene bands

wavenumber shifts observed with time are given in Table 2. It is apparent from Figure 3 that the bands initially observed from the pre-cursor polymer disappear after  $\sim T=5$  hours. However, after T=9 hours a weak feature is detected at 1255 cm<sup>-1</sup>. The constant wavenumber shift recorded for the  $\nu_2$  trans polyacetylene band, using  $\lambda_e$  457.9nm (see Table 2) is consistent with an isomerisation of short cis to short trans segments in the early part of the reaction. Indeed, it is only after several hours that a downward shift in the  $\nu_2$  band is observed, indicative of a significant change in the polyene chain length distribution.



The data obtained using red excitation (see Figure 4/Table 2) demonstrate the gradual movement of the 1261 and 1565 cm<sup>-1</sup> bands assigned to short <u>cis</u> segments to 1250 and 1541 cm<sup>-1</sup>, respectively, which previously have been assigned as the limits for an "infinitely" long <u>cis</u> segment.

TABLE 2 Raman v<sub>2</sub> wavenumber (cm<sup>-1</sup>) values of <u>cis</u> and <u>trans</u> polyacetylene isomers recorded at different stages of the reaction as it proceeded at room temperature

	λ <sub>e</sub> 457.9nm	λ <sub>e</sub> 647.1nm			
T/hrs	$v_2/cm^{-1}$ trans	ν <sub>2</sub> /cm <sup>-l</sup> cis	T/hrs	v2/cm <sup>-1</sup> trans	v <sub>2</sub> /cm <sup>-1</sup> cis
0	1538	1565(1260)	0	<del>-</del>	1565(1262)
1	1538	1565(1261)	1	-	1565(1260)
2	1538	-	2	-	1565(1261)
3	1538	-	3	_	1565(1262)
4	1537	-	4	-	1565(1260)
5	1537	-	5	1506	1564(1260)
9	1535	- (1255)	8	1495	1560(1257)
12	1535	- (1253)	10	1483	1552(1255)
24	1514(sh),1536	- (1252)	12	1483	1547(1254)
48	1507,1534(sh)	- (1251)	24	1479	1541(1251)
72	1503	1541(1251)	48	1479	1540(1251)
96	1501	1542(1251)	72	1478	-
			96	1478	

Values in parenthesis are for  $v_1$  <u>cis</u> isomer bands,  $v_2$  bands are lost under the intense  $v_2$  of the <u>trans</u> isomer when using  $\lambda_e$  457.9nm excitation. (sh) indicates shoulder.

It appears from these data, therefore, that the short <u>cis</u> segments observed initially gradually lengthen, as the elimination and transformation reaction proceeds, until they reach the limiting Raman values for an "infinitely" long chain.

Figure 4 also illustrates well the gradual build up of long trans polyacetylene chains. As can be seen, a very weak  $\nu_2$  band, at 1495 cm<sup>-1</sup>, is detected after  $\nu_3$  hours. The gradual downward shift of both the  $\nu_1$  and  $\nu_2$  bands and the increase in band

intensity is fully consistent with a change in the <u>trans</u> polyene sequence length distribution i.e. the concentration of long <u>trans</u> segments increases, as the reaction proceeds.

When monitoring the elimination and transformation reaction at 65 and 85°C, respectively, most of the subtle spectral changes, apparent when monitoring the reaction at room temperature, were missed. The gross changes were recorded and merit attention since information regarding the maximisation and elimination of the <u>cis</u> isomer can readily be obtained.

The most significant results obtained from the elevated temperature experiments relate to the final  $\nu_2$  values, i.e. 1496 cm<sup>-1</sup> ( $\lambda_e$  457.9nm) and 1472 cm<sup>-1</sup> ( $\lambda_e$  647.1nm), which were found to be invariant regardless of reaction temperature (up to 120°C).

In contrast to these values, the final  $v_2$  recorded by us from Shirakawa polymer has been 1498 cm<sup>-1</sup> ( $\lambda_e$  457.9nm) and 1460 cm<sup>-1</sup> ( $\lambda_e$  647.1nm). The similarity in  $v_2$  values recorded using blue excitation is not unexpected since both materials contain a certain amount of short chain length polymer. However, the difference in limiting  $v_2$  values using red excitation is thought to be very significant since, if the  $v_2$  values of 1460 cm<sup>-1</sup> corresponds to an "infinitely" long trans polyene chain, then it is thought that a value of 1472 cm<sup>-1</sup> corresponds to a shorter sequence length of > 30 carbon-carbon double bonds. This observation would be, in turn, consistent with the presence of some chemical or structural influence that limits the polyene chain length. Work is currently in progress to determine the exact nature of these defects.

(1) J.H. Edwards and W.J. Feast, Polymer Comm. 21, 595 (1980)

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